

Course Unit	Organic Chemistry			Field of study	Chemistry	
Bachelor in	Pharmacy			School	School of Health	
Academic Year	2022/2023	Year of study	1	Level	1-1	ECTS credits 5.0
Туре	Semestral	Semester	2	Code	9549-644-1206-00-22	
Workload (hours)	135	Contact hours	T - TP 2 T - Lectures; TP - Lectures a	2,5 PL 30 T	C - S - solving, project or laboratory; TC -	E - OT 7,5 O - Fieldwork; S - Seminar; E - Placement; OT - Tutorial; O - Other

Name(s) of lecturer(s)

Maria Alice Silva Pinto, Miguel José Rodrigues Vilas Boas

Learning outcomes and competences

- At the end of the course unit the learner is expected to be able to:

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 Plan and execute experiments in a laboratory of organic chemistry.
 Acquire the basic principles of chemical structure and bonding to understand the chemistry of the carbon compounds.
 Recognize the families of hydrocarbons and the main functional groups; Acquire and apply the concepts of structure and nomenclature of organic compounds.
 Recognize the aspects that determine the reactions characteristics in organic chemistry; Know the chemical and physical properties for the various families of organic the various families of organic compounds.
 Know the typical methods used for the preparation of the different organic compounds.
 Acquire the concepts to identify the spatial relationships between atoms and molecules.
 Apply the concepts of chemistry in health sciences.
 Implement and manane work independently and in groups.

- 8. Implement and manage work independently and in groups.

Prerequisites

- Before the course unit the learner is expected to be able to: 1. Recognize de structure and atomic properties.
- Nomenclature of inorganic compounds
 Fundaments of chemical equilibrium.

Course contents

Introduction and general principles of organic chemistry - Classification and nomenclature of organic compounds - Alkanes and cycloalkanes - Stereochemistry Alkenes - Alkines - Aromatic

Course contents (extended version)

- Introduction and general principles of organic chemistry

 Evolution of organic chemistry and its importance in health sciences.
 Structure and properties of carbon compounds Hybridization and geometry.
 The connections and interconnections in organic compounds: variability in physical properties.
 Isomery. Electronic effects: inductive and resonance effect.

 Classification and nomenclature of organic compounds.

 General aspects in the mechanisms of organic reactions.
 Alkanes and cycloalkanes.

- General aspects in the mechanisms of organic reactions.
 Alkanes and cycloalkanes
 Structure, conformations and physical properties.
 Synthesis : oxidation, pyrolysis and halogenation reactions.
 Stereochemistry.
- - The chirality in biological world. Enantiomers. Asymmetric carbon. Symmetry in achiral structures. Notation R and S. Physical properties of enantiomers. Molecules with two or more chiral centres. Resolution of enantiomers.
- 5. Alkenes

 - Structure, physical properties and conformations (isomers Cis/Trans and E/Z). Preparation of alkenes by elimination reactions: mechanisms E1 and E2. Addition reaction in alkene: hydration, halogenation, polymerization. Oxidation reactions: ozonolyse Alkenes of natural origin. Structure and classification of terpenes: beta-carotene and vitamin A.
- Alkenes of having origin: Substate and substation of experior of the perior of the peri
- 7. Aromatic compounds
 Structure and physical properties: the aromaticity.
- Electrophilic aromatic substitution reactions. Heteroaromatic hydrocarbons and polycyclic compounds.
 Contents for practice sessions
- Elemental analysis of organic compounds
 Synthesis and characterization of ethene

- Synthesis and characterization of entere
 Reactivity of aldehydes and ketones
 Synthesis and purification of an organic compound
 Meelting point evaluation
 Reactivity of alchools and esthers
- Recommended reading
- R. Morrison, R. Boyd; (2011). Química Orgânica (16ª Edição). Lisboa: Fundação Calouste Gulbenkian.
 T. W. G. Solomons, C. B. Fryhle, S. A. Snyder; (2018). Química Orgânica (12ª edição). Rio de Janeiro: LTC Livros Técnicos e Científicos Editora Lda.
 W. Brown, B. L. Iverson, E. Anslyn, C. S. Foote; (2018). Organic Chemistry, 8th Edition, Cengage Learning.
 D. L. Pavia, R. G. Engel, G. S. Kriz, G. M. Lampman; (2009), Química Orgânica Experimental, 2ª Edição, Bookman
 M. S. Heather, (2012). Basic Organic Chemistry for Allied Health Students. CreateSpace Independent Publishing Platform

- Teaching and learning methods

- Theory - One lesson per week of 2 hour. Interactive approach, using audiovisual materials. Study materials available via e-learning. - Practical classes - 1 lesson of 2 hours per week. Integration of knowledge with the resolution of nomenclature exercises. Performance of practical work, with educational and scientific laboratory equipment.

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Assessment methods

- Continous evaluation (Regular) (Final)

 Laboratory Work 12% (Student performance in the execution of pratical experiments and small quizs about the protocols.)
 Intermediate Written Test 9% (Written assay on the organic nomenclature)
 Intermediate Written Test 9% (Written assay on the results of practical experiments)
 Final Written Exam 70% (Written assay on the content of the theory-practical lessons.)

 Special evaluation (Regular) (Supplementary, Special)

 Final Written Exam 100% (This exam comprise questions about the practical experiments (30%) and the theoretical subject (70%))

 Working students (Student Worker) (Final, Supplementary, Special)

 Final Written Exam 100% (This exam comprise questions about the practical experiments (30%) and the theoretical subject (70%))

Language of instruction

Portuguese, with additional English support for foreign students.

Electronic validation			
Maria Alice Silva Pinto, Miguel José Rodrigues Vilas Boas	Olívia Rodrigues Pereira	Ana Maria Nunes Português Galvão	Adília Maria Pires da Silva Fernandes
03-03-2023	27-03-2023	28-03-2023	28-03-2023